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Archives of Applied Science Research, 2012, 4 (3):1359-1362
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Design, characterization and biological evaluation of a new series of *s*-triazines derived with quinolines

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ABSTRACT

4,6-dimethoxypyrimidin-2-amine condensed with trichloro *s*-triazine. The product of the above reaction was allowed to react with 8-Hydroxy quinoline. Finally various aromatic amines derivatives were allowed to react and the product were characterized by conventional and instrumental methods. Their structures were determined and important biological properties were studied.

Keywords: *s*-Triazine derivatives, 8-Hydroxy quinoline, Biological Evaluation.

INTRODUCTION

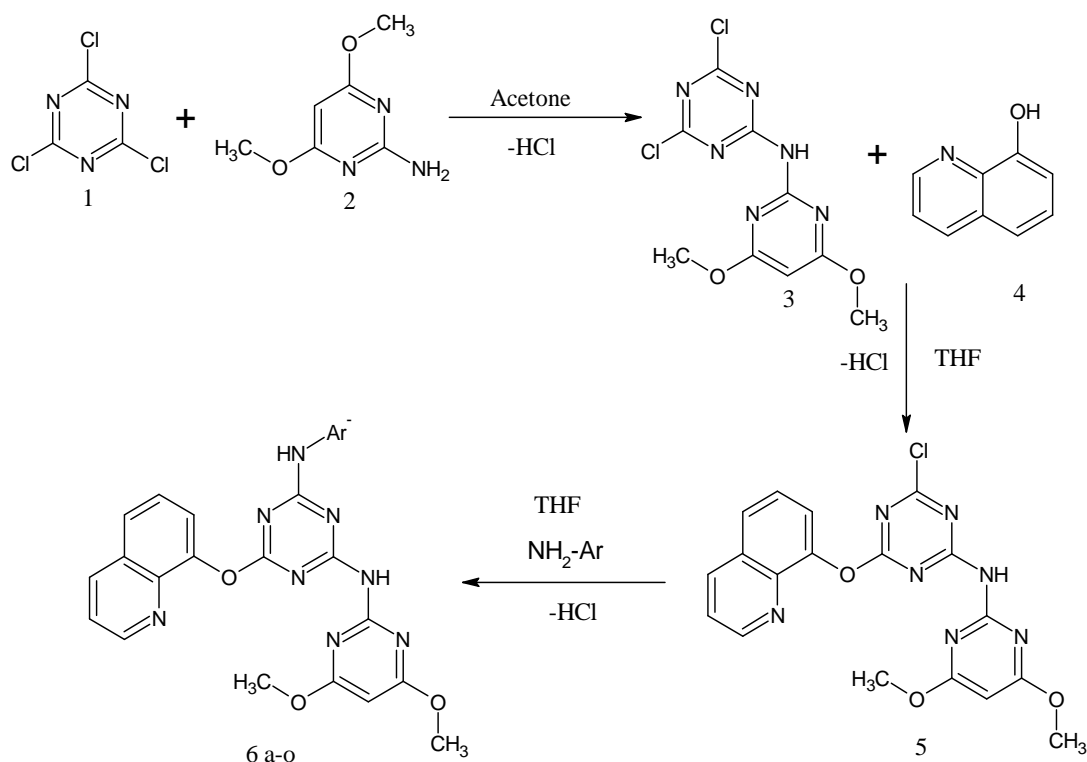
Nitrogen containing heterocycles play vital role in any industries. Among them 1,3,5-triazine represent a widely used lead structure with multitude of interesting application in numerous fields[1]. Several derivatives of *s*-triazine show antibacterial[2], antimicrobial[3] and herbicidal activities[4]. The replacement of a chlorine atom in cynuric chloride by basic group is greatly facilitated by the ring nitrogen atom of the symmetrically built *s*-triazine nucleus. 2,4,6-trichloro *s*-triazine derivatives prepared[5,6] by replacement of one chlorine atom at 0-5°C, second one at 35-45°C and third one at 80-100°C. Quinoline and their derivatives are receiving increasing importance due to their wide range of biological activities as antimalarial, antihypertensive, anti-inflammatory, antibacterial, antiasthmatic, antiplatelet activity and as tyrokinase inhibiting activity[7-10]. Pyrimidines and their derivatives possesses several interesting biological activities such as antimicrobial[11-15], antitumor[16] and antifungal activities[17]. Many pyrimidine derivatives are used for thyroid drugs and leukemia.

MATERIALS AND METHODS

The reagent grade chemicals were obtained from commercial sources and purified by either distillation or recrystallization before use. Purity of synthesized compounds has been checked by thin layer chromatography. Melting points were determined by open capillary method and are uncorrected. IR spectra are recorded on FT-IR Bruker with KBr disc. ¹H NMR spectra are recorded in DMSO-d₆ on a Bruker DRX-400 MHz using TMS as internal standard. The chemical shift are reported as parts per million(ppm) and mass spectra were determined on Jeol-SX-102(FAB) spectrometer.

Preparation of 4,6-dichloro-N-(4,6-dimethoxypyrimidin-2-yl)-1,3,5-triazin-2-amine

To a stirred solution of cyanuric chloride (0.01 mol) in acetone at low temperature, the solution of 4,6-dimethoxypyrimidin-2-amine (0.01 mol) in acetone was added and neutral P^H was maintained by adding 10% NaHCO₃ solution. The stirring was continued at the same temperature for 1 hours. Then stirring was stopped and solution was mixed with crushed ice. The product obtained was filtered and dried. The crude product was purified by recrystallization from Acetone to give 85% yield of the title compound. Melting Point 245°C.



Scheme-1

Synthesis route to *s*-triazine derivativesTable-1. Physical constants and elemental analysis of *s*-triazines.

C o m. no.	Ar-	Molecular Formula	M.P °C	Yield %	% of C Found, (calcd.)	% of H Found, (calcd.)	% of N Found, (calcd.)
6a	4-NO ₂ -C ₆ H ₄	C ₂₄ H ₁₉ N ₉ O ₅	272	80	56.15 (56.14)	3.74 (3.73)	24.56 (24.55)
6b	4-CH ₃ -C ₆ H ₄	C ₂₅ H ₂₂ N ₈ O ₃	280	89	62.24 (62.23)	4.62 (4.60)	23.24 (23.22)
6c	3,4-(Cl) ₂ -C ₆ H ₃	C ₂₄ H ₁₈ Cl ₂ N ₈ O ₃	289	85	53.65 (53.64)	3.39 (3.38)	20.86 (20.85)
6d	3-NO ₂ -C ₆ H ₄	C ₂₄ H ₁₉ N ₉ O ₅	272	79	56.15 (56.14)	3.74 (3.73)	24.57 (24.55)
6e	2-OH-4-NO ₂ -C ₆ H ₃	C ₂₄ H ₁₉ N ₉ O ₆	275	84	54.45 (54.44)	3.63 (3.62)	23.82 (23.81)
6f	2-OH-C ₆ H ₄	C ₂₄ H ₂₀ N ₈ O ₄	282	80	59.53 (59.50)	4.17 (4.16)	23.14 (23.13)
6g	2-C ₄ H ₃ N ₂	C ₂₂ H ₁₈ N ₁₀ O ₃	286	82	56.19 (56.17)	3.85 (3.86)	29.79 (29.77)
6h	2-Cl-C ₆ H ₄	C ₂₄ H ₁₉ ClN ₈ O ₃	279	80	57.33 (57.32)	3.82 (3.81)	22.29 (22.28)
6i	3-Cl-C ₆ H ₄	C ₂₄ H ₁₉ ClN ₈ O ₃	285	82	57.33 (57.32)	3.82 (3.81)	22.29 (22.28)
6j	2,4,5-(Cl) ₃ -C ₆ H ₂	C ₂₄ H ₁₇ Cl ₃ N ₈ O ₃	274	78	50.43 (50.41)	3.02 (3.00)	19.62 (19.60)
6k	2-OCH ₃ -C ₆ H ₄	C ₂₄ H ₂₀ N ₈ O ₄	278	75	59.52 (59.50)	4.17 (4.16)	23.14 (23.13)
6l	2,4-(NO ₂) ₂ -C ₆ H ₃	C ₂₄ H ₁₈ N ₁₀ O ₇	272	70	51.63 (51.62)	3.26 (3.25)	25.09 (25.08)
6m	2,4-(Cl) ₂ -2 NO ₂ -C ₆ H ₂	C ₂₄ H ₁₇ Cl ₂ N ₉ O ₅	282	79	49.52 (49.50)	2.95 (2.94)	21.66 (21.65)
6n	3-Cl-6-OH-C ₆ H ₃	C ₂₄ H ₁₉ ClN ₈ O ₄	271	83	55.56 (55.55)	3.68 (3.69)	21.57 (21.59)
6o	3-Cl-4-F-C ₆ H ₃	C ₂₄ H ₁₈ ClFN ₈ O ₃	277	77	55.35 (55.34)	3.49 (3.48)	21.53 (21.51)

Preparation of 4-chloro-N-(4,6-dimethoxypyrimidin-2-yl)-6-(quinolin-8-yloxy)-1,3,5-triazin-2-amine

To a stirred solution of 4,6-dichloro-N-(4,6-dimethoxypyrimidin-2-yl)-1,3,5-triazin-2-amine (0.01 mol) in THF at 35 °C the solution of 8-Hydroxy quinoline(0.01 mol) in THF was added drop wise and neutral P^H was maintained by 10% NaHCO₃ solution. The stirring was continued at the same temperature for three hours. Then stirring was stopped and solution was mixed with crushed ice. The product obtained was filtered and dried. The crude product was purified by recrystallization from THF. to give 75% yield of the title compound. Melting Point 302°C.

Preparation of N-(4-nitrophenyl)-N'-(4,6-dimethoxy pyrimidin-2-yl)-6-(quinolin-8-yloxy)-1,3,5-triazine-2,4-diamine

To the mixture of 4-chloro-N-(4,6-dimethoxypyrimidin-2-yl)-6-(quinolin-8-yloxy)-1,3,5-triazin-2-amine(0.01 mol) and 4-Nitro Aniline(0.01 mol) in THF was refluxed for 5-6 hours. The P^H was adjusted to neutral by adding 10% NaHCO₃ solution. After completion of reaction the content was added to cold water. The solid obtained was dried and crystallized from THF. Their physical constant data are given in Table-1 and synthetic scheme in Figure-1.

RESULTS AND DISCUSSION**Antimicrobial Activity***Antibacterial activity*

Antibacterial activity was carried out by broth dilution method [18]. The strains used for the activity were procured from Institute of Microbial Technology, Chandigarh. The compounds 6a-o were screened for their antibacterial activity against *Escherichia coli*, *Staphylococcus aureus*, *Pseudomonas aeruginosa* and *Staphylococcus pyogenes* of concentrations of 1000, 500, 200, 100, 50, 25, 12.5 µg/mL respectively (Table-2).

Antifungal activity

Same compounds were tested for antifungal activity against *C. albicans*, *A. niger* and *A. clavatus* of a concentrations of 1000, 500, 200, 100 µg/mL respectively(Table-2). The results are recorded in the form of primary and secondary screening. Each synthesized drug was diluted to obtain 1000 µg/mL concentration, as a stock solution.

Table 2. Antibacterial and Antifungal activities

Comp. No.	Minimal bactericidal concentration (MBC) in µg/ml				Minimal fungicidal concentration (MFC) in µg/mL		
	<i>E.coli</i>	<i>P.aerugi nosa</i>	<i>S. aureus</i>	<i>S.pyog enus</i>	<i>C. albicans</i>	<i>A.nigar</i>	<i>A.clavatus</i>
	MTCC -443	MTCC -1688	MTCC -96	MTCC -442	MTCC -227	MTCC -282	MTCC -1323
6a	100	200	500	500	250	250	250
6b	500	500	250	500	500	500	500
6c	250	250	500	250	500	500	500
6d	500	500	500	250	500	500	500
6e	250	250	250	500	500	500	500
6f	50	100	500	500	500	500	500
6g	250	500	250	250	500	500	500
6h	200	500	500	500	500	500	500
6i	50	200	500	500	1000	1000	1000
6j	100	250	500	250	500	500	500
6k	100	250	250	100	500	500	500
6l	50	100	1000	250	500	500	500
6m	500	500	1000	1000	500	500	500
6n	25	200	1000	1000	1000	1000	1000
6o	500	500	1000	500	500	500	500

For antibacterial activity, in present protocol 100 µg/mL is considered as moderately active, 50 µg/mL is considered as good activity and 25 µg/mL is considered as active as compared to the standard drug gentamycin. For antifungal activity 200 µg/mL is considered as moderately active, 100 µg/mL is considered as active as compared to standard drug Nystatin.

The synthesized drugs found to be active in this primary screening were further tested in a second set of dilution against all microorganisms. Secondary screening : The Drugs found active in primary screening were similarly diluted to obtain 100, 50, 25 µg/mL concentrations. 10 µL suspension from each well was further inoculated on appropriate media and growth was noted after 24 and 48 hrs. The lowest concentration, which showed no growth after spot subculture was considered as MBC/MFC for each drug. The highest dilution showing at least 99% inhibition was taken as MBC/MFC. The result of this test is affected by the size of the inoculums. The test mixture should contain 10⁸ organisms/mL. The standard drug used in the present study was "Gentamycin" for evaluating antibacterial activity which showed (0.25, 0.05, 0.5 & 1.0 µg/mL MBC against *S.aureus*, *E. Pyoganes* & *P. aeruginosa* respectively. "K. Nystation" was used as the standard drug for antifungal activity which showed 100 µg/mL MFC against fungi used for the antifungal activity. Compounds 6a, 6j and 6k were found to be moderately

active, 6f, 6i and 6l found to be active against *E.coli*, where as compound 6k found to be good active against *S. aureus*. This is due to the presence of chloro, methoxy, nitro, bromo, and hydroxyl in the *s*-triazine derivatives. Compound 6a was found to be moderately active against all the species of fungi.

Spectra study of *N*-(4-nitrophenyl)-*N'*-(4,6-dimethoxy pyrimidin-2-yl)-6-(quinolin-8-yloxy)-1,3,5-triazine-2,4-diamine

FT-IR (KBr) cm^{-1} : 3058(-N-H Str., Sec. amine), 1577(C=N Str., Sec. amine), 1498(C=N Str., ter. amine), 1363, 1400 (aromatic ring), 1255(C-O-C stretching), 802(disubstituted aromatic)

^1H NMR: 5.65 δ (s, C-NH-, 2H), 9.4 δ (s, C-NH-,1H), 6.6-8.738 (m, Ar-H, 10H), 8.07-8.09(d, 1H, -O-C=CH-CH=CH-), 8.14-8.19(t, 1H, -O-C=CH-CH=CH), 8.77-8.78(d, 1H, -O-C=CH-CH=).

MS: m/z. 514 with 78% relative intensity[M⁺].

Acknowledgments

We are grateful the SAIF, Punjab University for recording the spectra and Institute of Microcare Lab., Surat for providing Antimicrobial activities.

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