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# Synthesis and Effect of Eu Dopant on PL and Crystallites size of lanthanum Phosphate: LaPO<sub>4</sub>:Eu<sup>3+</sup>

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# ABSTRACT

In present paper the effect of dopant on the crystallite size, morphology and the photoluminescence of the sample were also investigated with X-ray spectrometer (XRD), Fourier-transform infrared spectroscopy (FTIR). The phosphors were synthesized using the standard solid state reaction technique and ground using mortar and pestle, fired at  $1200^{\circ}$ C for 1 hour in a muffle furnace. We have studied the effect of dopants on the Photo luminescent properties of the samples using Spectrofluorophotometer at room temperature. PL emission of undoped LaPO<sub>4</sub> phosphor was observed at 470 nm. Under the excitation of 254nm wavelength, PL emission of doped LaPO<sub>4</sub> phosphor shows peaks at 589, 596, 614 and 622nm with good intensity.

Keywords: Photoluminnescence, SEM ,XRD, Particcal size, rare earth ions, solid state reaction.

## INTRODUCTION

Recently there has been the great interest in the synthesis of various phosphors, because of the wide range of applications in optoelectronic fields. The morphology of the phosphor has an important effect on the emission intensity and the luminous intensity of the phosphors. The size and shape of the phosphors in growth becomes more and more important. Phosphors are widely used in displays and lighting devices. Over the past a few years, they have been applied in many fields, such as optical display panels, cathode ray tubes, optoelectronic, sensitive device, nanoscale electronic and plasma display panels[2–4]because of their excellent performance high density excitation and high energy quantum excitation. Various solution-phase routes, including solid state reaction, sol-gel, precipitation, water oil micro emulsion, polyol-mediated process, ultrasonification, hydrothermal, and mechano chemical method[6-9], have been tried to lower the reaction temperature and obtain high-quality LaPO<sub>4</sub> based nano particles. However, the simple and mass fabrication of LaPO<sub>4</sub> nanocrystals with narrow grain size distribution and uniform morphology still remains a challenge. Eu3+ doped LaPO4 is a red phosphor widely used in fluorescent lamps due to its high quantum efficiency.[2] We adopted the standard solid state reaction technique to prepare LaPO<sub>4</sub> with good morphologies and fine crystal structures; and its emission and intensity of luminescence were also studied.



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#### MATERIALS AND METHODS

The base sample LaPO4 and LaPO<sub>4</sub> phosphor doped with Eu rare-earth ions, were prepared using solid state synthesis method. Stoichiometric proportions of raw materials namely, Lanthanum Oxide (La<sub>2</sub>O<sub>3</sub>), Diammonium Hydrogen Phosphate [(NH<sub>4</sub>)<sub>2</sub> H PO<sub>4</sub>), Europium Oxide (Eu<sub>2</sub>O<sub>3</sub>) were grinded in an agate motor and mixed and compressed into a crucible and heated at  $1200^{\circ}$ C for 1 hour in a muffle furnace. The prepared samples were again powdered for taking the measurements.

The XRD patterns of the samples were obtained using Diffractometer system XPERT-PRO at NCL Pune and the excitation & emission spectra were recorded at room temperature using (SHIMADZU,make Spectroflurophotometer RF – 5301 PC) using Xenon lamp as excitation source at display research Lab., Department of Applied Physics, M.S.U. Baroda.. The emission and excitation slit were kept at 1.5 nm.FT-IR measurements were carried out on BRUKER Make,Vector-22 spectrometer in transmittance mode. With KBr tablet cell as the background spectrum, spectra were collected in a vacuous environment at a resolution of 1 cm<sup>-1</sup> over the wave number of 4000-400 cm<sup>-1</sup>. To record IR spectra, about 100 mg of the mixture of sample and pure and dry potassium bromide is grounded and heated .Then mixture was compressed into a sample cell at the pressure 12 ton for 1 min .



#### **RESULTS AND DISCUSSION**

The crystallinity and phase purity of the product were firstly examined by XRD analysis. Fig 1 and 2 for pure LaPO<sub>4</sub> and LaPO<sub>4</sub> dopped Eu reply shows the typical X-ray diffraction (XRD) patterns of synthesized samples. As shown XRD patterns of nanocrystals are in good agreement in the values from JCPDS no.32-0493 of LaPO<sub>4</sub>, which shows that all the products are monazite LaPO<sub>4</sub> with monoclinic structure. This implies that the La ions in LaPO<sub>4</sub> system are replaced by Eu ions with smaller atomic radius. The diffraction patterns were obtained using CuK<sub>α</sub> radiation ( $\lambda = 1.540598 \text{ A}^0$ ) at 40 kv and 30 mA, and divergence slit fixed at 1.52 mm. Measurements were made from  $2\theta = 10^0$  to  $80^0$  with steps of 0.008356<sup>0</sup>.

When crystallites are less than approximately 100 nm in size, appreciable broadening in X-ray diffraction lines occurs. The grain size of particles of powder sample were calculated by using Scherer equation

$$D=0.9 \lambda / \beta \cos\theta$$

Where  $\beta$  represents full width at half maximum of XRD lines.

The main peak was found around 28.7<sup>°</sup> corresponding to a d- value of about 3.12A<sup>°</sup>, followed by other less intense peaks corresponds to the monoclinic system of crystal structure of Lanthanum Phosphate.

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758



The crystallite size of LaPO<sub>4</sub> is 59 nm. The crystallite size increases to 76 nm in LaPO4 doped with Eu<sup>3+</sup>

#### **Energy Dispersive X-Ray Analysis (EDAX):**

An elemental analysis of base sample lanthanum phosphor (LaPO4) was carried out by employing the energy dispersive X-ray analysis technique which provides local information of concentration of different elements.Fig.3 shows the EDAX spectra of LaPO4 in which the presence of La, O, and P are clearly identified from table 1. Here the element N is may be present due to the atmospheric nitrogen.



Fig.3. EDAX spectra of LaPO<sub>4</sub>

## Fourier transform infrared spectroscopy(FT-IR):

The technique has been used to identify the reaction between solids, by monitoring the vibrational and rotational motion of the molecules during the reaction. The FTIR spectrum of LaPO4 and LaPO4 doped Eu heated at  $1200^{\circ}$ C has been depicted in fig .4 and fig.5 resply.

#### Table 1. Quantitative analysis of LaPO<sub>4</sub>

El AN Series unn. C norm. C Atom. C Error [wt.%] [wt.%] [at.%] [%]	
La 57 L-series 46.58 45.14 9.09 O 8 K-series 42.24 40.93 71.54 P 15 K-series 7.96 7.71 6.96 N 7 K-series 6.41 6.22 12.41	7.4 0.4

Total: 103.19 100.00 100.00



Fig.4.FT-IR spectra of LaPO4



Fig.5.FT-IR specra of LaPO4:Eu

The data of infra red spectra of LaPO<sub>4</sub> in fig.4 presents the characteristic bands of the rare earth phosphate (assigned to the  $PO_4^{3-}$  group) in the 500-1100 cm<sup>-1</sup> wave number range. The wave numbers are:1091, 1059, 992, 952, 771,

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# Y. S. Patil et al

617, 576,563,539 cm<sup>-1</sup>. The bands observed at 1271,1400 ,3134 cm<sup>-1</sup> can be attributed to the presence of water (stretching vibration of the O-H bond i.e. hydroxyl complexes) adsorbed by KBr during the pellet formation. The band observed at 3134 cm<sup>-1</sup> may be due to the stretching vibration of O-H and N-H associated bonds. This band is not present in the sample LaPO<sub>4</sub> doped Eu. The bands located at 1271 and 1400 cm<sup>-1</sup> is attributed to N-O bond in NO<sup>3-</sup> i.e. assigned to anti-symmetry stretching vibration of NO<sup>3-</sup>.

The data from fig.5 presents the characteristic bands of the rare earth phosphate (assigned to the  $PO_4^{3-}$  group) in the 500-1100 cm<sup>-1</sup> wave number range. The wave numbers are:1096,1091,992,952,771,618,577,563,538 cm<sup>-1</sup>. The peaks found in v<sub>3</sub> region are: 952,992,1091 cm<sup>-1</sup> and in v<sub>4</sub> region are : 538,563,577,618 cm<sup>-1</sup> in the pattern indicates that the specimen is LaPO<sub>4</sub>:Eu<sup>3+</sup> with typical monoclinic structure.

Scanning Electron Microscope(SEM)



Fig.6.SEM image of LaPO<sub>4</sub>



Fig.7.SEM image of LaP: Eu



Fig 6 and fig.7 shows the SEM images of pure LaPO4and LaPO4:Eu sintered at  $1200^{0}$ Cfor 4 hours appears unuiform may be due to the formation of fractal attribution to sort of self organization. The particles having irregular shapes of an average basal diameter of 300 nm and length of 1.5  $\mu$ m.

#### **Photoluminescence study**:

The fig.8 and fig.9 gives the excitation and emission spectra of pure LaPO<sub>4</sub>.

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The PL emission of undopped  $LaPO_4$  phosphor was observed at 470 nm under the excitation of 254 nm wavelength a perfect blue region with very good intensity.

The emission spectra of LaPO4: Eu (0.5%) is shown in fig.10 reveals the modification of the emission wavelength of pure phosphor. The peaks at 589 nm and 594 nm corresponding to orange red colour are derived from the allowed magnetic dipole moment transition  $[{}^{5}D_{0} \rightarrow {}^{7}F_{1}]$ , whose intensity is affected by the crystal environment surrounding Eu. The peaks at 614 and 622 nm corresponding to red colour are generated from the forced electric transition  $[{}^{5}D_{0} \rightarrow {}^{7}F_{1}]$ . The Eu ions allow to occupy a site without inversion center. The intensities  ${}^{5}D_{0} \rightarrow {}^{7}F_{3}$  and  ${}^{5}D_{0} \rightarrow {}^{7}F_{4}$  are highly suppressed as compared to the intensified  ${}^{5}D_{0} - {}^{7}F_{2}$ 



#### CONCLUSION

The phase purity has been verified by XRD, SEM, EDAX and FTIR spectral analysis. The XRD data indicates that the peak position do not change with the substitution of La by Eu into monazite type  $LaPO_4$  lattice. The PL intensity is very high therefore  $LaPO_4$ : Eu phosphors can be easily applied in various types of lamp and display.

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