Available online at <u>www.scholarsresearchlibrary.com</u>



Scholars Research Library

Der Pharmacia Lettre, 2015, 7 (2):206-220 (http://scholarsresearchlibrary.com/archive.html)



Synthesis of some new compounds containing the pyrazolyl moiety

Galal Hosni Sayed^a, Sara Nabil Abd El-al^b and Zeinab Hashem^b

^aHeterocyclic Synthetic Laboratory, Department of Chemistry, Faculty of Science, Ain Shams University, Abbassia, Cairo, Egypt ^bHeterocyclic Synthetic Laboratory, Department of Chemistry, Faculty of Science (Girls), Al-Azhar University, Nasr

City, Cairo, Egypt

ABSTRACT

The one pot reaction of pyrazolone3 with malononitrile, ethylacetoacetate in presence of aromatic aldehyde, ammonium acetate and/or piperidine gave compounds 4-6. Reaction of compound 4 with ammonium thiocaynate, formamide / formic acid and acetic anhydride gave compounds 7-9. Reaction of compound 5 with acetic anhydride followed by hydrazine hydrate gave compounds 10 and 11 respectively. Reaction of pyrazolone 3 with glucose, phenyl isothiocaynate, nitrous acid /aniline, aromatic aldehyde have also been taken onto consideration to give compounds 12,13,15 and the a β –unsaturated ketone 16. Compound 16 was used as key intermediate for preparation of various heterocyclic compounds via its reaction with pyrazolone3, hydrazine hydrate, hydroxylamine hydrochloride, urea, thiourea, ethylcyanoacetate / sodium ethoxide, malononitrile / ammonium acetate or sodium ethoxide and also with ethyl acetoacetate / sodium ethoxide. The newly synthesized compounds were characterized by IR, ¹HNMR, mass spectral data. The antitumor and biological activity has also been taken in consideration.

Keywords: Pyrazolone ,Pyrazolopyrimidines,Pyrazolopyrazole, Pyranopyrazole , Biological activity , , Antitumor activity.

INTRODUCTION

Pyrazolone is one of the important heterocyclic compounds having five membered ring lactam and an additionalketo (C=O) group . It occurs in many drugs and synthetic products. The compounds having pyrazolone moiety are found to be remarkable anti tubercular, anti-inflammatory, antibacterial, and antitumor activity^[1-8] as potential therapeutics for immune thrombocytopenias⁹ bacteriostatic¹⁰⁻¹⁵ anticancerous¹⁶, antioxidant¹⁷⁻²⁰ and fungicidal^{21,22}.

The present investigation deals with synthesis of some new pyrazolone derivatives to study their behavior towards some nucleophilic as well as electrophilic reagents in the hope of obtaining new derivatives of biological interest and potential target compounds. The new synthesized pyrazolone were tested for their cytotoxic activity and their structure activity relationship were examined

Galal Hosni Sayed et al

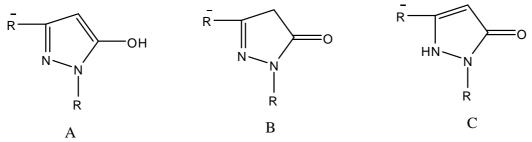
MATERIALS AND METHODS

General

All melting points were measured on a GallenKamp melting point apparatus and are uncorrected. IR spectra (KBr) were recorded with a Perkin Elmer Spectrum RXIFT-IR system. ¹HNMR(DMSO-D6) were measured with a Varian Gemini 200 MHz instrument using TMS as internal standard and DMSO-d6 as solvent and mass spectra were measured with a Shimadzu GC–MS–QP 100 EX mass spectrometer.

RESULTS AND DISCUSSION

The new derivatives were prepared following the reaction sequences depicted in Scheme 1 and 2 The starting material pyrazolone3 was prepared using ethyl acetoacetate 1 as β -keto ester and 2,4- dinitro phenyl hydrazine 2. Compounds unsubstituted at pyrazole C-4, three isomers are possible assigned as 1H-pyrazol – 5-ol (A), 2,4-dihydro -3H-pyrazol-3-one(B) and 1,2 –dihydro -3H-pyrazol-3-one(C).



The IR spectrum of **3** displayed band at 1658 cm⁻¹ due to C=O absorption at position 5 and was devoid for OH and NH which indicated that pyrazolone**3** exists in the keto form.

The bifunctional compound **3** was used as key intermediate for the preparation of several new heterocyclic compounds via its multicomponent reaction with p-hydroxybenzaldehyde and active methylene compounds namely malononitrile and / or ethyl acetoacetate in the presence of ammonium acetate and / or piperidine to give the aminocarbonitrile**4** and **5** and the pyrazolopyridine **6**.

It has been reported thataminocarbonitrile reacted with ammonium thiocyanate in boiling acetic acid to give (thioxopyrimidinyl) thiourea derivative , while in our investigation we isolated the amino (iminomethyl) thiocyanatsmethane thioimidate7 through two consecutive addition of two molecules of ammonium thiocyanate to compound **4**.

Treatment of compound 4 with formamide in the presence of formic acid and DMF afforded the open structure pyrazolo(3,4-b) pyridinylformamide $\mathbf{8}$.

Acetylation of compounds4 and 5 with acetic anhydride afforded the acetylated compounds pyrazolo (3,4-b-pyridn-6-yl) acetamide 9 and dihydropyrano (2,3-c) pyrazol-6-yl) acetamide10, respectively. The structure of these compounds were confirmed by spectroscopic data and also the reaction of compound 10 with hydrazine hydrate to give the bis compound 11. The hydrazine hydrate is an ambiendt nucleophile therefore one molecule of it attacks two molecules of compound 10 followed by hydrogenation to afford the bis compound 11.

Condensation of pyrazolone 3 with glucose in ethanol / DMF / drops of AcOH mixture gave the pentahydroxyhexylidenepyrazolone derivative 12 .

In the presence of piperidine ,pyrazolone 3 reacted with phenyl isothiocyanate through its active methylene group to give the thioamide derivative 13.

Compound 13 is a versatile multifunctional reagent which reacted with hydrazine hydrate to affordpyrazolo (3,4-c) pyrazolethione 14 .

Reaction of pyrazolone 3 with nitrous acid and aniline gave the red dye 4-(2-phenylhydrazono)-1H-pyrazolone derivative 15.

Knoevenagel condensation reaction of the pyrazolone 3 with 2,4-dimethoxybenzaldehyde in the presence of piperidine gave the pyrazolone derivative 16.

Pyrazolone 16 reacted with pyrazolone 3 in the presence of sodium ethoxide to give thehydroxypyrazolylpyrazolol17.

Pyrazolone 16 can be used as key intermediate for the preparation of several new heterocyclic compounds, via its reaction with different nitrogen nucleophiles such as hydrazine hydrate , hydroxyl amine hydrochloride , urea and thiourea (Scheme 2,3)

Reaction of compound 16 with hydrazine hydrate gavepyrazolo (3,4-c) pyrazole 18 via β -attack on C=C moiety in compound 16 followed by 1,5- dipolar cyclization , while hydroxylamine hydrochloride gave pyrazolo(3,4-c) isoxazole 19.

Reaction of 16 with urea afforded the pyrazolo(3,4-c) pyrimidinone 20 via β -attack on the C=C moiety in compound 16, followed by 1,6- intramolecular dipolar cyclization.

However, reaction of thioureawith compound 16 gave the pyrazolo (4,3-e)(1,3) thiazinylthiourea 21 via β -attack on the C=C moiety in compound 16, followed by 1,6- intramolecular dipolar cyclization through the sulfur atom, then the loss of ammonia through the reaction with another molecule of thiourea.

Reaction of ethyl cyanoacetate in the presence of sodium ethoxide gave the pyrano (2,3-c)pyrazolecarbonitrile 22.

Interestingly, the product of the reaction of malononitrile with compound 16 depends on the reaction conditions, in the presence of ammonium acetate and / or sodium ethoxide compounds 23 and 24 were obtained, respectively.

Similarly ,reaction of compound 16 with ethyl acetoacetate in the presence of ammonium acetate gave thepyrazolo(3,4-b) pyridinylethanone 25 via the addition of the nucleophile ethyl acetoacetate on the β - carbon of the electrophile 16 in a conjugate addition reaction in the presence of ammonium acetate followed by cyclization and dehydrogenation

DISCUSSION

Synthesis of 1-(2,4-dinitrophenyl)-3-methyl-1H-pyrazol-5(4H)-one (3)

A mixture of ethylacetoacetate 1 (0.01 mol) and 2,4-dinitrophenylhydrazine 2 (0.01 mol) in acetic acid (50ml) was refluxed for 4 hrs and left to cool. The separated solid was filtered off, washed with ethanol, dried and recrystallized from aceticacid.yield 85% and M.P168 $^{\circ}$ C.

Analysis of **3**C₁₀H₈N₄O₅(264) (%) calcdC: 45.46; H, 3.05; N, 21.21.Found:C, 45.40; H, 3.09; N, 21.22.

The IR spectrum of 3showed absorption bands at 1658 cm⁻¹ and 1608 cm⁻¹ attributable to $\sqrt{C=O}$ and $\sqrt{C=N}$.

Synthesis of 6-amino-1-(2,4-dinitrophenyl)-4-(4-hydroxyphenyl)-3-methyl-1H-pyrazolo[3,4-b]pyridine-5carbonitrile (4)

A mixture of pyrazolone3(0.01 mol), malononitrile(0.01 mol), ammonium acetate (0.03mol) and p-hydroxybenzaldehyde(0.01 mol)wasfused for 15 hrs at 140-170 $^{\circ}$ C, the reaction mixture was poured onto water to give the solid product 4 ,which was washed with water dried and recrystallized from EtOH .yield 72% and M.P250 $^{\circ}$ C.

Analysis of 4C₂₀H₁₃N₇O₅(431) (%) calcd: C, 55.69; H, 3.04; N, 22.73. Found: C, 55.40; H, 3.29; N, 22.82.

IR spectrum of 4showed absorption bands at 1623 cm⁻¹, 2236 cm⁻¹, 3107 cm⁻¹, 3334,3226 cm⁻¹ attributable to $\sqrt{C=N}$, \sqrt{CN} and \sqrt{OH} and $\sqrt{NH_2}$. The mass spectrum showed the molecular ion peak at m/z 432 (M+1,0.03%). The ¹H-

NMR(DMSO-d6) showed signal bands δppm at 9.89 (s,1H,OH), 6.91-7.91 (m, 7H,Ar-H), 4.84(s,2H,NH₂), 2.34(s,3H, CH₃).

Synthesis of 6-amino-1-(2,4-dinitrophenyl)-4-(4-hydroxyphenyl)-3-methyl-1,4-dihydropyrano[2,3-c]pyrazole-5-carbonitrile (5)

A mixture of pyrazolone 3(0.01 mol), malononitrile(0.01 mol), and p-hydroxybenzaldehyde(0.01 mol) in presence of 1 ml piperdinewas fused for 15 hrs at 160 °C. After cooling, the solid product was washed with EtOH, dried and recrystallized from EtOH. yield 68% and M.P280 °C.

Analysis of **5**C₂₀H₁₄N₆O₆(434) (%) calcd: C, 55.30; H, 3.25; N, 19.35..Found:C, 55.40; H, 3.20; N, 19.30.

IR spectrum of 4showed absorption bands at 1613 cm⁻¹, 2233 cm⁻¹, 3345,3289 cm⁻¹ attributable to $\sqrt{C=N}$, \sqrt{CN} and $\sqrt{NH_2}$. The mass spectrum showed the molecular ion peak at m/z 434 (0.1%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 10.29 (s,1H,OH), 6.78-8.01 (m, 7H,Ar-H), 4.22(s,2H,NH₂), 1.39(s,3H, CH₃).

Synthesis of 4-(1-(2,4-dinitrophenyl)-3,6-dimethyl-1H-pyrazolo[3,4-b]pyridin-4-yl)phenol(6)

A mixture of pyrazolone3(0.01 mol), ethylacetoacetate (0.01 mol), ammonium acetate (0.03mol) and p-hydroxybenzaldehyde(0.01 mol)wasfused for 15 hrs at 170 $^{\circ}$ C. After cooling, the solid product was washed withwater, filtered off and recrystallized from EtOH. yield74% andM.P278 $^{\circ}$ C.

Analysis of $5C_{20}H_{15}N_5O_5(405)$ (%) calcd: C, 59.26; H, 3.73; N, 17.28. Found: C, 59.40; H, 3.69; N, 17.18.

IR spectrum of **6**showed absorption bands at 1618 cm⁻¹, 3356cm⁻¹ attributable to $\sqrt{C=N}$, \sqrt{OH} . The mass spectrum showed the molecular ion peak at m/z 407 (26.62%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 6.96-8.11 (m, 8H,Ar-H), 4.01(s,2H,NH₂), 2.19(s,6H,2X CH₃).

Synthesis of 6-amino-1-(2,4-dinitrophenyl)-4-(4-hydroxyphenyl)-3-methyl-1H-pyrazolo[3,4-b]pyridin-5-yl)(iminomethyl)thiocyanatomethane thioimidate(7)

A mixture of 4 (0.01 mol) and ammonium thiocyanate (0.03mol) in acetic acid (10 ml) was refluxed for 10 hrs. The reaction mixture was poured onto water ,filtered off , washed with water , dried and recrystallized from EtOH.yield68% and M.P266 $^{\circ}$ C.

Analysis of $7C_{22}H_{15}N_9O_5S_2$ (549) (%) calcd:C, 48.08; H, 2.75; N, 22.94; S, 11.67. Found: C, 48.01; H, 2.77; N, 22.96; S, 11.70.

IR spectrum of **7**showed absorption bands at 1616 cm⁻¹, 2206 cm⁻¹,3069cm⁻¹,3354cm⁻¹attributable to $\sqrt{C=N}$, \sqrt{CN} , \sqrt{OH} , \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 547 (3.32%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 6.76-8.91 (m, 7H,Ar-H),5.11(s,2H,2XNH),4.51(s,2H,NH₂), 3.19(s,3H, CH₃).

Synthesis of N-(5-cyano-1-(2,4-dinitrophenyl)-4-(4-hydroxyphenyl)-3-methyl-1H-pyrazolo[3,4-b]pyridin-6-yl)formamide (8)

Amixture of **4** (0.01 mol), formic acid(5 ml), formamide(10 ml) and dimethylformamide (5 ml) was refluxed for 2 hrs, the reaction mixture was poured ontowater. The solid product that separated was filtered off, washed well with water, dried and recrystallized from EtOH.yield79% and M.P273 $^{\circ}$ C.

Analysis of**8** C₂₁H₁₃N₇O₆(459) (%) calcd: C, 54.91; H, 2.85; N, 21.34 .Found: C, 54.88; H, 2.83; N, 21.37.

IR spectrum of **8**showed absorption bands at 1612 cm⁻¹, 1657 cm⁻¹, 2206 cm⁻¹, 3069cm⁻¹, 3461cm⁻¹ attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{CN} , \sqrt{OH} , \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 459 (1.02%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 11.45(s,1H,NH) ,10.91(s,1H,OH), 7.06-8.51 (m, 8H,Ar-H),), 2.34(s,3H, CH₃).

Synthesis of N-(5-cyano-1-(2,4-dinitrophenyl)-4-(4-hydroxyphenyl)-3-methyl-1H-pyrazolo[3,4-b]pyridin-6-yl) acetamide (9), N-(5-cyano-1-(2,4-dinitrophenyl)-4-(4-hydroxyphenyl)-3-methyl-1,4-dihydropyrano[2,3-c] pyrazol-6-yl)acetamide(10)

Amixture of compound **4 and/or 5** (0.01 mol), and acetic anhydride(20 ml was refluxed for 5 hrs ,the reaction mixture was poured onto water. The solid product that separated was filtered off, washed well with water, dried and recrystallized from EtOH.Yield of **97**2% and M.P286°C.Yield of **10**75% and M.P298°C.

Analysis of $P_{22}H_{15}N_7O_6(473)$ (%) calcd: C, 55.82; H, 3.19; N, 20.71; Found: C, 55.85; H, 3.17; N, 20.68; while for $10C_{22}H_{16}N_6O_7(476)$ (%) calcd: C, 55.47; H, 3.39; N, 17.64; Found: C, 55.40; H, 3.42; N, 17.70.

IR spectrum of **9**showed absorption bands at 1613 cm⁻¹, 1675 cm⁻¹, 2206 cm⁻¹, 3355 cm⁻¹attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{CN} , $\sqrt{OH/\sqrt{NH}}$ while for **10**showed absorption bands 1595 cm⁻¹, 1666 cm⁻¹, 2206 cm⁻¹, 3356 cm⁻¹attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{CN} , $\sqrt{OH/\sqrt{NH}}$. The mass spectrum of **9** showed the molecular ion peak at m/z 473 (10.55%) and for **10** is 477(2.88%). The ¹H-NMR(DMSO-d6) of **9** showed signal bands δ ppm at 11.65(s,1H,NH), 10.89(s,1H,OH), 7.01-8.01 (m, 7H,Ar-H),), 2.28(s,6H,2x CH_3), while **10**showed signal bands δ ppm at 11.74(s,1H,NH), 10.95(s,1H,OH), 7.03-8.41 (m, 8H,Ar-H),), 1.98(s,6H,2x CH_3)

Synthesis of 6,6⁻-(1,1-hydrazine-1,2-diyl)bis(ethane-1,1-diyl)bis(azanedinyl)bis (1-(2,4-dinitrophenyl)-4-(4-hydroxyphenyl)-3-methyl-1,4-dihydropyrano[2,3-c]pyrazole-5- carbonitrile(11)

Amixture of **10** (0.01 mol), hydrazine hydrate(0.01 mol)in ethanol (50 ml) was refluxed for 2 hrs ,the reaction mixture was poured ontoice/HCl(1:3). The solid product that separated was filtered off, washed well with water, dried and recrystallized from EtOH.yield72% and M.P 212 $^{\circ}$ C.

Analysis of**8** C₄₄H₃₆N₁₄O₁₂(951) (%) calcd: C, 55.46; H, 3.78; N, 20.59 .Found: C, 55.48; H, 3.77; N, 20.58.

IR spectrum of **11** showed absorption bands at 1599 cm⁻¹, 2216 cm⁻¹, 3226cm⁻¹, 3425cm⁻¹ attributable to $\sqrt{C=N}$, \sqrt{CN} , \sqrt{OH} , \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 950 (2.04%).

Synthesis of 1-(2,4-dinitrophenyl)-3-methyl-4-(2,3,4,5,6-pentahydroxyhexylidene)-1H-pyrazol-5(4H)-one (12) A mixture of **3** (0.01 mol), glucose (0.01 mol)in water (1 ml),ethanol(50 ml), and dimethylformamide(5 ml) was heated on water bath for 2 hrs.The solid product that separated was filtered off, washed well with water(3x10 ml), dried and recrystallized from EtOH.yield69% and M.P 232 °C.

Analysis of **12** C₁₆H₁₈N₄O₁₀(426) (%) calcd: C, 45.08; H, 4.26; N, 13.14.Found: C, 45.01; H, 4.30; N, 13.17.

IR spectrum of **12**showed absorption bands at1613cm⁻¹, 1698 cm⁻¹, 3341cm⁻¹attributable to $\sqrt{C=N}$, $\sqrt{C=O\sqrt{OH}}$. The mass spectrum showed the molecular ion peak at m/z 426 (55.55%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 11.45(s,1H,NH) ,10.98(s,5H,5xOH), 7.36-8.11 (m, 4H,Ar-H),), 3.34(s,4H, 4xCHOH),3.21(s,2H,CH₂OH),1,99(s,3H,CH₃).

Synthesis of 1-(2,4-dinitrophenyl)-3-methyl-5-oxo-N-phenyl-4,5-dihydro-1H-pyrazole-4-carbothioamide (13) A mixture of **3** (0.01 mol), and phenylisothiocyanate(0.01 mol) in dry acetone (50 ml) in presence of piperidine (0.5 ml) was heated on water bath for 4 hrs .The reaction mixture was poured ontoice/HCl.The solid product that separated was filtered off, washed well with water, dried and recrystallized from EtOH.yield81% and M.P 289 °C.

Analysis of **13** $C_{17}H_{13}N_5O_5S(399)$ (%) calcd: C, 51.12; H, 3.28; N, 17.54; S, 8.03. Found: C, 51.09; H, 3.30; N, 17.55; S, 8.03.

IR spectrum of **13**showed absorption bands at 1613 cm⁻¹, 1674 cm⁻¹, 3169cm⁻¹,attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{OH} . The mass spectrum showed the molecular ion peak at m/z 400 (0.09%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 7.12-8.01 (m, 8H,Ar-H),4.45(s,1H,NH),2.55(s,1H.CH), 1.34(s,3H, CH₃).

Synthesis of 6-(2,4-dinitrophenyl)-4-methyl-2-phenyl-1,6-dihydropyrazolo[3,4-c]pyrazole-3(6H)-thione (14) Amixture of 13 (0.01 mol), and hydrazine hydrate(0.01 mol) in ethanol (50 ml) was refluxed for 5 hrs. The reaction mixture was poured ontoice/HCl. The solid product that separated was filtered off, washed well with water, dried and recrystallized from EtOH.yield58% and M.P330 °C.

Analysis of 14 $C_{17}H_{12}N_6O_4S(396)$ (%) calcd: C, 51.51; H, 3.05; N, 21.20; S, 8.09.Found C, 51.45; H, 3.07; N, 21.22; S, 8.11.

IR spectrum of **14**showed absorption bands at 1436 cm⁻¹, 1613 cm⁻¹, 3226cm⁻¹,attributable to $\sqrt{C=S}$, $\sqrt{C=N}$, \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 398 (4.09%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 6.92-8.11 (m, 8H,Ar-H),4.23(s,1H,NH),1.83(s,3H, CH₃).

Synthesis of 1-(2,4-dinitrophenyl)-3-methyl-4-(2-phenylhydrazono)-1H-pyrazol-5(4H)-one (15)

Aniline(0.01 mol) was dissolved in con. HCl (5ml) and the solution was then cooled to 0.5° C .Solution of sodium nitrite (0.01 mol) in water (3 ml) was then added to this solution dropwise with vigorous stirring during about 1 hr ,while cooling at 0.5° C . The clear diazonium salt solution was then added dropwise to a well cooled(0.5° C) and stirred solution of pyrazolone**3** (0.01 mol) in sodium acetate (1 gm,dissolved in 5 ml of 25 % aqueows ethanol). Stirring was continued for 4 hrs atto 0.5° C. The solid product that separated was filtered off , washed well with water , dried and recrystallized from EtOH.yield71% and M.P221 °C.

Analysis of **15** C₁₆H₁₂N₆O₅(368) (%) calcd: C, 52.18; H, 3.28; N, 22.82; FoundC, 52.22; H, 3.26; N, 22.80.

IR spectrum of **15**showed absorption bands at 1608 cm⁻¹, 1663 cm⁻¹, 3226cm⁻¹, attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 370 (11.01%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 11.23(s,1H,NH),6.82-8.51 (m, 8H,Ar-H),2.13(s,3H, CH₃).

Synthesis of 4-(2,4-dimethoxybenzylidene)-1-(2,4-dinitrophenyl)-3-methyl-1H-pyrazol-5(4H)-one (16)

Amixture of **3** (0.01 mol), 2,4- dimethoxybenzaldehyde(0.01 mol) in ethanol (50 ml) in presence of piperidine (1 ml) was refluxed for 5 hrs. The solid product that separated was filtered off, washed well with ethanol, dried and recrystallized from EtOH.yield74% and M.P197 $^{\circ}$ C.

Analysis of **16** C₁₉H₁₆N₄O₇(412) (%) calcd: C, 55.34; H, 3.91; N, 13.59.FoundC, 55.30; H, 3.93; N, 13.61.

IR spectrum of **16**showed absorption bands at 1598 cm⁻¹, 1673 cm⁻¹, attributable to $\sqrt{C=N}$, $\sqrt{C=O}$. The mass spectrum showed the molecular ion peak at m/z 414 (14.19%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 7.12-8.31 (m, 7H,Ar-H),3.23(s,6H,2x OCH₃),1.99(s,3H, CH₃).

Synthesis of 4,4'-((2,4-dimethoxyphenyl)methylene)bis(1-(2,4-dimitrophenyl)-3-methyl-1H-pyrazol-5-ol) (17) Compound 16 (0.01 mol) was added at room temperature with stirring to a mixture of pyrazolone3 (0.01 mol) ,and sodium ethoxide (0.5 gm sodium in 10 ml ethanol), then the mixture was stirred for another 1 hr at 100° C. The reaction mixture was poured ontoice/HCl.The solid product that separated was filtered off, washed well with water , dried and recrystallized from EtOH.yield74% and M.P190 °C.

Analysis of **17** C₂₉H₂₄N₈O₁₂(676) (%) calcd: C, 51.48; H, 3.58; N, 16.56; Found C, 51.52; H, 3.56; N, 16.54.

IR spectrum of **17**showed absorption bands at 1625 cm⁻¹, 3340cm⁻¹, attributable to $\sqrt{C=N}$, \sqrt{OH} . The mass spectrum showed the molecular ion peak at m/z 676 (11.21%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 11.23(s,2H,2xOH),6.87-8.02 (m, 10H,Ar-H),3.41(s,6H,2x OCH₃), 2.13(s,6H,2X CH₃).

Synthesis of4-(2,4-dimethoxyphenyl)-1-(2,4-dimitrophenyl)-3-methyl-1,3a,4,5-tetrahydropyrazolo[3,4-c] pyrazole(18)

Amixture of 16 (0.01 mol), hydrazine hydrate(0.01 mol) in ethanol (20 ml) was refluxed for 5 hrs. The solid product that separated after cooling was filtered off, washed well with water, dried and recrystallized from EtOH.yield84% and M.P270 $^{\circ}$ C.

Analysis of **18** C₁₉H₁₈N₆O₆ (426)(%) calcd: C, 53.52; H, 4.26; N, 19.71. Found: C, 53.55; H, 4.25; N, 19.69.

IR spectrum of **18**showed absorption bands at 1599 cm⁻¹, 3226 cm⁻¹, attributable to $\sqrt{C=N}$, \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 425 (10.12%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 11.34(s,1H,NH), 7.01-8.11 (m, 8H,Ar-H),3.41(s,6H,2x OCH₃),2.01(s,3H, CH₃).

Synthesis of c]isoxazole(19) 3-(2,4-dimethoxyphenyl)-6-(2,4-dimitrophenyl)-4-methyl-3a,6-dihydro-3H-pyrazolo[3,4-

Amixture of **16** (0.01 mol), hydroxylamine hydrochloride(0.01 mol) in pyridine (10 ml) was refluxed for 6 hrs. The solid product that separated after cooling was filtered off, washed well with water, dried and recrystallized from EtOH.yield59% and M.P230 $^{\circ}$ C.

Analysis of **19**C₁₉H₁₇N₅O₇ (427)(%) calcd: C, 53.40; H, 4.01; N, 16.39. FoundC, 53.45; H, 3.98; N, 16.36.

IR spectrum of **19**showed absorption bands at 1614 cm⁻¹, attributable to $\sqrt{C=N}$, The mass spectrum showed the molecular ion peak at m/z 425 (13.02%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 11.34(s,1H,NH), 7.21-8.01 (m, 7H,Ar-H),3.65(s,1H,CH)3.21(s,6H,2x OCH₃),2.21(s,3H, CH₃).

Synthesis of 4-(2,4-dimethoxyphenyl)-1-(2,4-dinitrophenyl)-3-methyl-4,5-dihydro-1H-pyrazolo[3,4-d] pyrimidin -6(3aH)-one (20)

Amixture of **16** (0.01 mol), and urea (0.01 mol) in glacial acetic acid (20 ml) was refluxed for 5hrs .The solid product that separated after cooling was filtered off, washed well with water, dried and recrystallized from EtOH.yield77% and M.P214 $^{\circ}$ C.

Analysis of **20**C₂₀H₁₈N₆O₇ (454)(%) calcd: C, 52.86; H, 3.99; N, 18.50. FoundC, 52.90; H, 3.97; N, 18.548.

IR spectrum of **20**showed absorption bands at1600cm⁻¹,1711 cm⁻¹,3363 cm⁻¹ attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 454 (1.02%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 11.04(s,1H,NH), 7.01-8.31 (m, 7H,Ar-H),3.45(s,1H,CH)3.01(s,6H,2x OCH₃),2.09(s,3H, CH₃).

Synthesis of1-(4-(2,4-dimethoxyphenyl)-1-(2,4-dinitrophenyl)-3-methyl-1,4-dihydropyrazolo[4,3-e][1,3]thiazin -6-yl)thiourea (21)

Amixture of 16 (0.01 mol), and thiourea(0.01 mol) in glacial acetic acid (20 ml) was refluxed for 5hrs . The solid product that separated after cooling was filtered off, was hed well with water, dried and recrystallized from EtOH. yield61% and M.P241 °C.

Analysis of $21C_{20}H_{18}N_6O_6$ (529)(%) calcd: C, 47.63; H, 3.62; N, 18.52; S, 12.11.FoundC, 47.58; H, 3.64; N, 18.53; S, 6.83.S, 12.12.

IR spectrum of **21**showed absorption bands at 1421cm⁻¹,1615cm⁻¹, 3110-3001 cm⁻¹·3279 cm⁻¹ attributable to $\sqrt{C=S}$, $\sqrt{C=N}$, $\sqrt{NH_2}$

Synthesisof 4-(2,4-dimethoxyphenyl)-1-(2,4-dinitrophenyl)-3-methyl-6-oxo-1,6-dihydropyrano[2,3-c]pyrazole-5-carbonitrile (22)

Amixture of **16** (0.01 mol), and ethylcyanoacetate(0.01 mol) in sodium ethoxide (0.5 gm in20 ml) was refluxed for 5hrs. The solid product that separated after cooling and pouring into ice was filtered off, washed well with water, dried and recrystallized from EtOH.yield57% and M.P256 $^{\circ}$ C.

Analysis of **22**C₂₂H₁₅N₅O₈ (477)(%) calcd: C, 55.35; H, 3.17; N, 14.67.FoundC, 55.30; H, 3.20; N, 14.69.

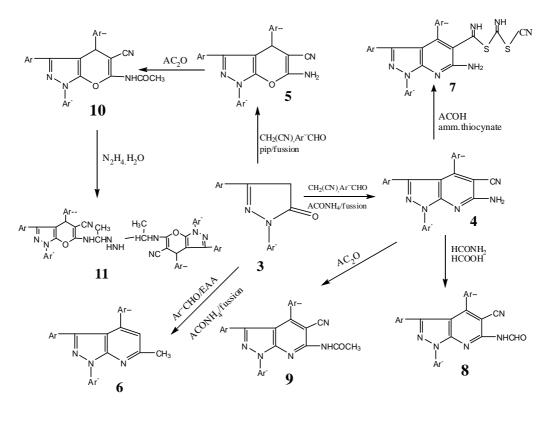
IR spectrum of **22**showed absorption bands at 1617cm⁻¹,1709cm⁻¹, 2220 cm⁻¹ attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{CN} . The mass spectrum showed the molecular ion peak at m/z 479 (5.02%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at 6.89-8.14 (m, 8H,Ar-H),3.09 (s,6H,2x OCH₃),1.87(s,3H, CH₃).

Synthesisof 2-[5-amino-4-(2,4-dimethoxyphenyl)-1-(2,4-dinitrophenyl)-3-methyl-4-9-dihydro-1H-pyrazolo[4⁻,3⁻:5,6]pyrido(2,3-d)pyrimidin-7-yl]acetonitril(23)

Amixture of **16** (0.01 mol) ,malononitrile(0.01 mol) and ammonium acetate(2 gm) was fused for 5 hrs at 170° C. After cooling the solid washed well with water ,filtered off , dried and recrystallized from EtOH.yield63% and M.P270 °C.

Analysis of 23C25H21N9O6 (543)(%) calcd: C, 55.25; H, 3.89; N, 23.19.FoundC, 55.20; H, 3.91; N, 23.22.

IR spectrum of **23**showed absorption bands at 1598cm⁻¹, 2234 cm⁻¹, 3398attributable to $\sqrt{C=N}$, \sqrt{CN} , \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 545 (4.02%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at ,11.42(s,1H,NH),6.69-8.01 (m, 7H,Ar-H),4.23(s,2H,NH₂),3.43 (s,6H,2x OCH₃),2.11(s,3H, CH₃).



 $Ar=CH_3$ $Ar'=_{2,2}4-(NO_2)_2C_6H_3$ $Ar''=4-(OH)C_6H_4$

Scheme 1

Synthesis of 2-[5-amino-4-((2,4-dimethoxyphenyl)-1-(2,4-dinitrophenyl)-3-methyl1,4-dihydropyrazolo[4⁺,3⁺:5,6]pyrano [2,3-d]pyrimidin-7-yl]acetonitrile (24)

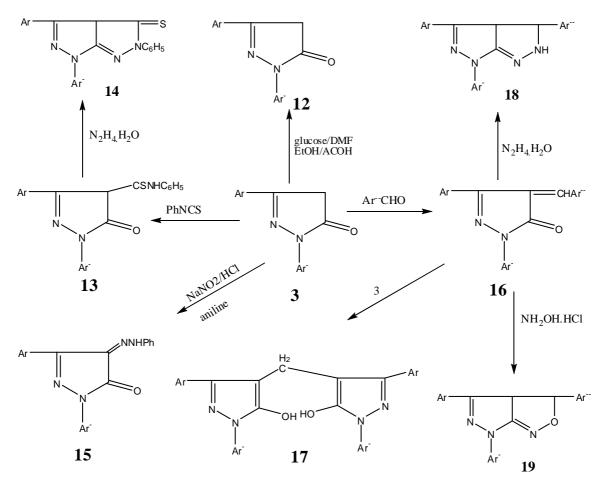
Amixture of **16** (0.01 mol) ,malononitrile(0.01 mol) and sodium ethoxide(0.5 gm in 20 ml) was refluxed for 6 hrs. The mixture was poured onto ice /HCl. The solid product that separated was filtered off, washed well with water, dried and recrystallized from EtOH.yield54% and M.P232 $^{\circ}$ C.

Analysis of $24C_{25}H_{20}N_8O_7(544)(\%)$ calcd: C, 55.15; H, 3.70; N, 20.58.FoundC, 55.10; H, 3.72; N, 20.61. IR spectrum of 24showed absorption bands at 1598cm⁻¹, 2234 cm⁻¹ 3398attributable to $\sqrt{C=N}$, \sqrt{CN} , \sqrt{NH} . The mass spectrum showed the molecular ion peak at m/z 545 (4.02%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at ,11.42(s,1H,NH),6.69-8.01 (m, 7H,Ar-H),4.23(s,2H,NH₂),3.43 (s,6H,2x OCH₃),2.11(s,3H, CH₃).

Synthesis of 4-(2,4-dimethoxyphenyl)-1-(2,4-dinitrophenyl)-6-hydroxy-3-methyl-1,3a,4,7a-tetrahydro pyrano [2,3-c]pyrazol-5-yl)ethanone(25)

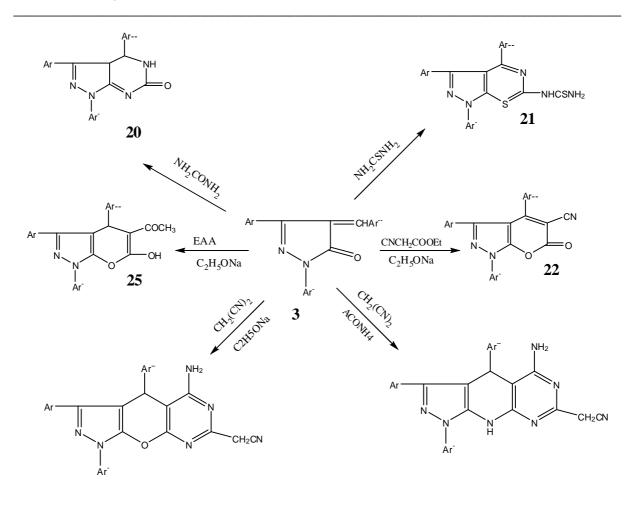
Amixture of **16** (0.01 mol) ,ethylacetoacetate(0.01 mol) and sodium ethoxide(0.5 gm in 20 ml) was refluxed for 6 hrs. The mixture was poured onto ice /HCl. The solid product that separated was filtered off, washed well with water, dried and recrystallized from EtOH.yield73% and M.P214°C.

Analysis of $25C_{23}H_{22}N_4O_9(496)(\%)$ calcd: C, 55.65; H, 4.06; N, 11.29.FoundC, 55.50; H, 4.16; N, 11.33. IR spectrum of 25showed absorption bands at 1613cm⁻¹, 1675 cm⁻¹·3355attributable to $\sqrt{C=N}$, $\sqrt{C=O}$, \sqrt{OH} . The mass spectrum showed the molecular ion peak at m/z 498 (7.03%). The ¹H-NMR(DMSO-d6) showed signal bands δ ppm at ,10.42(s,1H,OH),6.99-8.06 (m, 7H,Ar-H),3.13 (s,6H,2x OCH₃),2.35(s,3H,COCH₃),1.91(s,3H, CH₃).



Ar=CH₃ Ar⁻=
$$_{2,4-(NO_2)_2C_6H_3}$$
 Ar⁻= 2,4-(OCH₃)₂C₆H₃

Scheme 2



Ar=CH₃ Ar⁻= $_{2,4-(NO_2)_2C_6H_3}$ Ar⁻= 2,4-(OCH₃)₂C₆H₃

Scheme 3

4. Cytotoxic assay

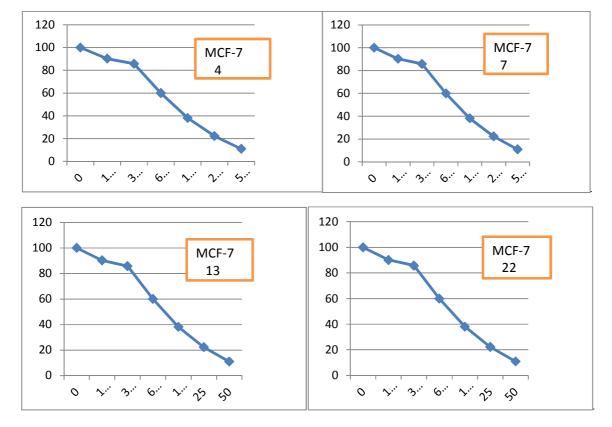
Cytotoxic assay was performed using the modified method as previously described (Tengchaisir.T,1998). Cancer cells were grown in Ham's/F12 medium containing 2 m Ml-glutamine supplemented with 100 U/ml penicillin, streptomycin and 10% FBS.

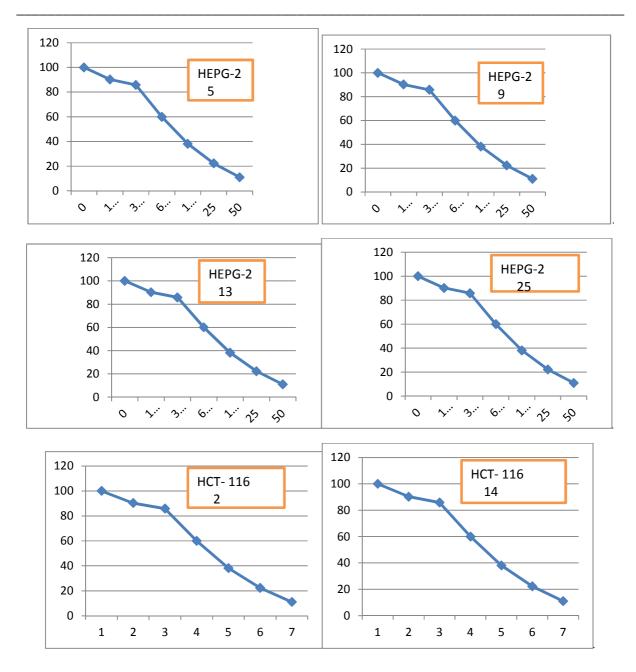
Except Hep G-2 cell was grown in DMEM. Briefly , cell lines(Table 1) suspended in RPMI- 1640 containing 10% FBD were seeded at 1x104 cells(100 μ L) per well ina96—well plate , and incubated in humidifiedatmosphere , 95% air 5% CO2 at 37 °C . After 24h, additional medium (100 μ L) containing the tested compound and vehicle was added to a final concentration of 50 μ g/ml , 0.2% DMSO , and further incubated for 3 days. Cells were subsequently fixed with 95% EtOH, stained with crystal violet solution, and lysed with a solution of 0.1 N HCl in MeOH after which absorbance was measured at 550 nm whereas Hep G2, HCT and MCF-7 cells were stained by MTT.

Conc.µg/ml		M	CF7			HE	PG2		HCT116				
	4	7	13	22	5	9	16	25	2	14	19	23	
0	100	100	100	100	100	100	100	100	100	100	100	100	
1.56	74.24	69.01	88.98	79.00	87.00	95.65	97.57	89.92	82.32	76.03	90.00	90.23	
3.125	31.53	33.98	65.16	72.48	66.42	81.17	91.42	76.56	60.07	55.18	71.60	85.81	
6.25	22.16	21.43	48.54	56.77	59.14	60.86	80.14	54.09	35.10	40.22	59.89	59.98	
12.5	10.01	15.87	43.67	34.53	49.68	37.98	60.03	22.67	19.90	19.25	26.80	38.14	
25	8.11	10.05	33.42	22.65	11.47	22.43	30.14	11.21	12.65	10.02	14.94	22.27	
50	3.35	5.14	15.67	2.66	4.19	11.06	7.10	4.98	5.44	2.03	6.98	10.98	

Table(1):Effect of some new prepared compounds on different types of tumor cells as cytotoxic drug

IC50 values were determined as the drug and sample concentrations at 50% inhibition ofcell growth (Fig.1)





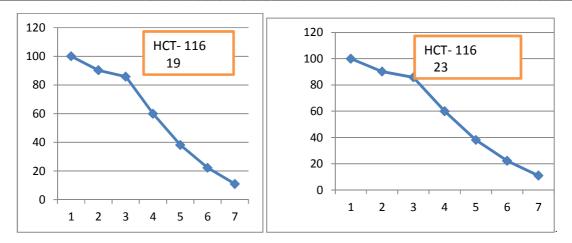


Table (2): Cytotoxic activity of pyrazolyl derivatives (2, 4, 5, 6, 7, 9, 13, 14, 16, 19, 22, 23, 25)

Cell lines ^a	IC50 (µg/ml) ^{b,c}											
	2 4 5 7 9 13 14 16 19 22 23 2									25		
Hep G-2	NT	NT	4.19	NT	11.06	NT	NT	7.10	NT	NT	NT	4.98
НСТ	5.44	NT	NT	NT	NT	NT	2.03	NT	6.98	NT	10.98	NT
MCF-7	NT	3.35	NT	5.14	NT	15.67	NT	NT	NT	2.66	NT	NT
NT: indicates not tested												

^a Cancer cell lines were hepatocellular carcinoma cell line (Hep G-2); colon carcinoma cell line (HCT); breast carcinoma cell line (MCF-7). ^bWhen IC50 > 50 μg/ml denotes inactive compound. ^c The assays were performed in triplicate.

5. Cytotoxic activity:

Cytotoxic activity of pyrazolyl analogs (2,4,5,7,9,13,14,16,19,22,23,25) against three cancer lines using a modified method.

The results which were listed in (Table 2) showed that compounds **2,4,5,7,14,22** and **25** have shown the highest activity toward the tested cancer cell lines (Hep G-2)(MCF-7) and (HCT) respectively with IC50 of 5.44,3.35,4.19,5.14,2.66,2.03, µg/ml and 4.98µg/ml respectively.Fig (**2**).

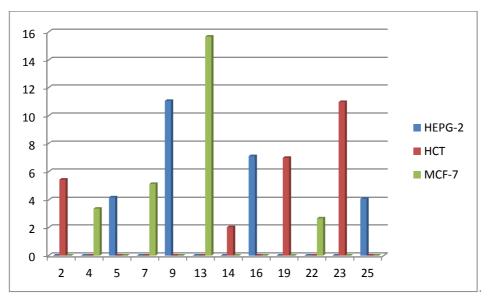


Figure (2): Cytotoxic activity of some pyrazole derivatives

6-Antimicrobial activity

Antimicrobial activity of pyrazolyl analogs (3,6,15,20) against three different gram +ve and gram -ve showed significant activity.

Samples	3			6			15			20		
Con.	1	5	2.5	1	5	2.5	1	5	2.5	1	5	2.5
Aspergillusfungatus	0	++	+	0	+	0	+++	0	0	0	0	++
Penicilliumitalicum	+++	0	+	0	+	0	0	0	0	++	0	+
Syncephala strum racemosum	0	0	0	+++	0	0	+	0	0	0	0	++
Candida albicans	0	++	0	0	+	0	+++	0	0	0	0	0
Staphylococcus aureus	0	0	+++	+	0	0	0	+	0	0	+	0
Pseudomonas aeruginosa	+++	0	0	0	+	0	0	++	++	0	0	0
Bacillus sbutilis	0	0	0	++	0	+	0	0	0	++	+	0
Escherichia coli	+	++	0	0	+	0	0	0	+	0	0	++

Table(3):antimicrobial activity

 $Inhibition \ values = 0.1-0.5 \ cm \ beyond \ control = +; \ Inhibition \ values = 0.6-1.0 \ cm \ beyond \ control = ++; \ Inhibition \ values = 1.1-1.5 \ cm \ beyond \ control = +++; \ 0 = not \ tested$

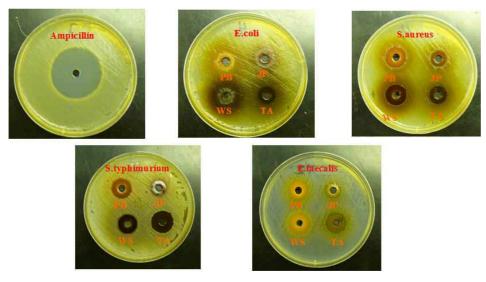


Figure (3): inhibition zone of some pyrazole derivatives

REFERENCES

[1] Castagnolo, D.; Manetti, F.; Radi, M.; Bechi, B.; Pagano, M.; De Logu, A.; Meleddu, R.; Saddi, M.; Botta, M. *Bioorg. Med. Chem.* **2009**, *17*, 5716–5721.

[2] Radi, M.; Bernardo, V.; Bechi, B.; Castagnolo, D. Tetrahedron Lett. 2009, 50, 6572–6575.

[3] Singh, Da.; Singh, De. J. Indian Chem. Soc. 1991, 68, 165–167.

[4] W. Wilson and N. Bottiglieri, *Cancer Chemo-Therapy*, 137. *International Journal of Organic and Bioorganic Chemistry* **2014**; 4(1): 10-13

[5] Moreau, F.; Desroy, N.; Genevard, J.M.; Vongsouthi, V.; Gerusz, V.; Le Fralliec, G.; Oliveira, C.; Floquet,S.; Denis, A.; Escaich, S.; Wolf, K.; Busemann, M.; Aschenbrenner, A. Discovery of new Gram-negativeantivirulence drugs: Structure and properties of novel E. coli WaaC inhibitors. *Bioorg. Med. Chem. Lett.***2008**, *18*, 4022–4026.

[6] Alessandro Balbia, Maria Anzaldia, Chiara Macciòa, CinziaAiellob, Mauro Mazzeia, Rosaria Gangemi^b, PatrizioCastagnolac, MariangelaMielea, CamilloRosanod, Maurizio Viale, *Eur. Jor.Med. Chem.*, 2014, 46(11), 5293–5309

[7] Tantawy, A.; Eisa, H.; Ismail, A.; Alexandria, M. E. J. Pharm. Sci. 1988, 2, 113–116.

[8] Pasha, F.A.; Muddassar, M.; Neaz, M.M.; Cho, S.J. J. Mol. Graph. Model. 2009, 28, 54-61.

[9] Anita Dubey, Govt. M.L.B. College Bhopal, M.P. India, Journal of Natural Sciences Research. 2014, 4(6) 125-129.

[10] HavvaBalseven, M. Mustafa İşgör, SametMert, ZuhalAlım, ŞükrüBeydemir, Salim Ok, RahmiKasımoğulları, *Bioorganic & Medicinal Chemistry*, **2014**, *22* (*9*)*2739-2752*.

[11] Bhatti H, Seshadri S: Color Technol, 2004, 120:151–155.

[12] Jae-Hong C., Sung-Hee H., Eui-Jae L., Andrew D., J Soc Dyers Colourists., 115, 32 (1999).

[13] K.Hunger, Industrial Dyes, Chemistry, Properties, Applications. Wiley-VCH Weinheim., 20 2003.

[14] M. A. Metwally, Yaser. A. M. Solomon et al International Journal of Modern Organic Chemistry, 2012, 1(3) 213-225.

[15] W. Wilson and N. Bottiglieri, *Cancer Chemo-Therapy*, 137. *International Journal of Organic and Bioorganic Chemistry* **2014**; 4(1): 10-13

[16] G.S. Sahariya and H.R. Sharma, J. Indian Chem. Soc., 51(1974) 351.

[17] AjayaKabra, G.S. Sahariya and H.R. Sharma, J. Indian Chem. Soc., 52 (1975) 989

[18] P.S. Fernandes, (Miss) BhateSandhya, (Miss) Philip Gupta and V.V. Nandkarni, J. Indian Chem. Soc., 52 (1976) 830.

[19] S.K. Mohanti, R. Shridhar, S.Y. Padmanavan ,SundarRao and A.S. Mittra, *Indian J. Chem.*, 15B (**1977**) 1146. [20] Towns A: *Dyes Pigm*, **1999**, 42:3–282

[21] Peng-Cheng Lv, Huan-Qiu Li, Juan Sun, Yang Zhou, Hai-Liang Zhu, Bioorganic & Medicinal Chemistry, 18, I 13, 1 2010, 4606-4614

[22] AnuradhaDoma, RavindraKulkarni, RadhakrishnaPalakodety, G. NarahariSastry, JanardhanSridhara, AchaiahGarlapati, *Bioorganic & Medicinal Chemistry*, 22, 21, 1 **2014**, 6209-6219.